MIGRATION OF METHYL GROUP IN HEPTAMETHYLBENZENONIUM ION A NEW ROUTE TO PERDEUTEROHEXAMETHYLBENZENE

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Completely deuterated hexamethylbenzene was obtained earlier by isotopic exchange reaction with liquid deuterated ammonia in presence of potassium amide (1).

We offer a new route to perdeuterohexamethylbenzene via heptamethylbenzenonium ion (II) shown by Doering et al.(2) to be formed as a result of protonation of 4-methylene-1,1, 2,3,5,6-hexamethylcyclohexadiene-2,5(I).

Since this reaction is reversible the incorporation of deuterium into 4-CH₃ group must occur in deuteroacids.

We have recently found all the methyl n.m.r. peaks of II in trifluoroacetic acid to collapse to a single line at elevated temperatures. This has been explained by the rapid 1536 No. 21

intramolecular migration of methyl group in II (3). The combination of migration with isotopic exchange reaction of hydrogen in 4-CH₃ group affords the possibility of replacing all hydrogen atoms of II by deuterium. The elimination of one of geminal CD₃ groups leads to perdeuterohexamethylbenzene.

A solution of I(605 mg.) in 9 ml. of deuterotrifluoroacetic acid was held at 70° for 15 min. under a dry nitrogen
atmosphere and the most part of acid (ca.90%) was removed in
vacuo at room temperature. After 6-fold repetition of this procedure 2 ml. of deuterotrifluoroacetic acid and 2 ml. of
deuterium oxide were added to the residue (1.8 g.) and the
solution was held at 70° for 48 hrs. The solid was separated
and discolved in benzene, and the solution was washed with
concentrated hydrochloric acid and water. The evaporation
of benzene yielded 425 mg. of solid, m.p.165°(EtOH), lit.(1)
m.p. 165,5°. Mass spectral analysis showed the total content
of deuterium in hexamethylbenzene obtained to be 98 atom %.
Infrared spectrum: V_{max}^{KS} . 2250 (s), 2200 (s), 2187 (sh), 2173
(sh), 2117 (s), 2075 (vs), 1630 (m), 1412 (vs), 1093 (sh),
1047 (s), 900 (m), 675 (w) cm⁻¹.

When carrying out the isotopic exchange reaction of hydrogen in II in the presence of hexamethylbenzene the latter incorporated no deuterium. This confirms the intramolecular mechanism of migration of the methyl group in II.

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